



ALEXANDER ROAD HIGH SCHOOL
PHYSICAL SCIENCE CONTROL TEST
GRADE 11

MARCH 2014
IC, CO, KB

1 HOUR
TOTAL = 60

Instructions

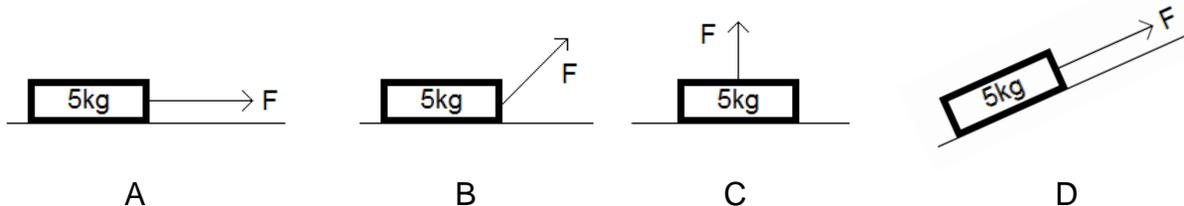
- The question paper consists of 5 questions.
- Answer all the questions.
- Answer section A on the answer sheet provided AND section B on folio sheets.
- Rule off after each question in Section B.
- A non-programmable calculator may be used.
- Number the answers correctly according to the numbering system
- Round off to two (2) decimal places where necessary.
- A periodic table has been included on the back of the answer sheet

SECTION A

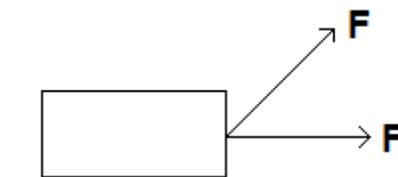
- Answer on the answer sheet -

QUESTION 1: Multiple choice

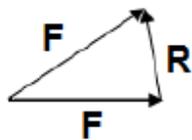
1.1 A 5kg block is placed on a surface and pulled with a 5N force (F). In which case will the normal force be the **GREATEST**...



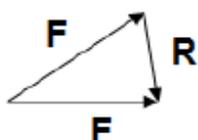
1.2 A crate is pulled along a frictionless surface by two forces, each with a magnitude of F, as shown in the diagram...



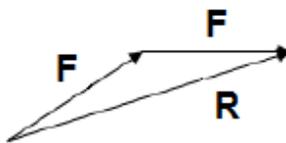
Which vector diagram correctly shows how the resultant force R on the crate can be determined?



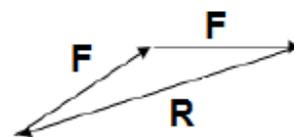
A



B



C



D

1.3 Who established the following relationship?

The pressure exerted on an enclosed gas in a set volume is directly proportional to the temperature.

- A Boyle
- B Charles-Lussac
- C Charles
- D Gay-Lussac

1.4 A chemical reaction with an enthalpy change of $-26\text{kJ}\cdot\text{mol}^{-1}$ is a(n) _____ reaction.

- A Exothermic
- B activated
- C complex
- D Endothermic

1.5 Which of the following molecules has the highest bond energy?

- | | |
|----------------|----------------|
| A H_2 | B F_2 |
| C O_2 | D N_2 |

1.6 The angles between the bonds in a tetrahedral molecule are...

- | | |
|-----------------|-----------------|
| A 120° | B $109,5^\circ$ |
| C $104,5^\circ$ | D 90° |

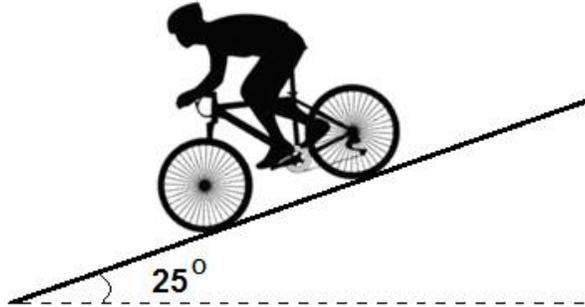
[6 x 2 = 12]

SUB - TOTAL: 12

SECTION B

QUESTION 2

A cyclist free wheels (ie. moves without pedaling) down a straight stretch of road that makes an angle of 25° with the horizontal. The mass of the cyclist and his bike is 100kg. Throughout the motion, the cyclist experiences a frictional force of 120N. The cyclist and his bike are to be considered as a single object.



- 2.1 Draw a free-body diagram showing all the forces acting on the object. (3)
- 2.2 Determine the components of the objects weight, F_g :
- 2.2.1 Acting parallel to the slope, $F_{g//}$ (2)
- 2.2.2 Acting perpendicular to the slope, $F_{g\perp}$ (2)
- 2.3 If the kinetic frictional force is 120N calculate:
- 2.3.1 The magnitude of the resultant force acting parallel to the slope. (2)
- 2.3.2 The magnitude of the normal force. (1)
- 2.3.3 The coefficient of kinetic friction. (2)

[12]

QUESTION 3

A passenger in a motor car notices that an air-freshener hanging from the rearview mirror of the car looks like it moves backwards when the car accelerates forward away from a stop street.

- 3.1 State Newton's first law of motion. (2)
- 3.2 Use it to explain this observation. (2)

[4]

QUESTION 4

- 4.1.1 Draw the Lewis structure for NH₃. (2)
- 4.1.2 How many lone pairs and bonding pairs are present in the NH₃ molecule? (2)
- 4.1.3 What is the VSEPR name given to the shape of this molecule (NH₃). (1)
- 4.1.4 Are the bonds in the NH₃ molecule polar or non-polar?
(Use calculations to support your answer) (3)
- 4.1.5 Using a sketch, indicate whether the NH₃ molecule is polar or non-polar? (2)
- 4.1.6 Explain why the NH₃ molecule has a different shape to the BF₃ molecule. (3)
- 4.1.7 Use Lewis Structures to show the formation of the NH₄⁺ ion through dative covalent bonding. (3)

[16]

QUESTION 5

- 5.1 500 cm³ of helium gas is kept in a container at a pressure of 90kPa. If the temperature remains constant, calculate:
- 5.1.1 The volume of the helium at a pressure of 100kPa. (3)
- 5.1.2 Calculate the amount of moles of gas present at 27°C. (4)
- 5.2 Define absolute temperature. (2)
- 5.3 Draw a graph of volume versus temperature and **indicate** on the graph where and how absolute zero temperature was graphically and theoretically established. (4)
- 5.4 Draw a labeled graph of E_p vs *Progress of reaction* for an endothermic reaction (3)

[16]

TOTAL 60 MARKS

Gas Laws:

$$\frac{P_1V_1}{T_1} = \frac{P_2V_2}{T_2}$$

$$PV = nRT$$

$$n = \frac{m}{M}$$

Mechanics: friction: $f_s = \mu_s N$ $f_k = \mu_k N$