



ALEXANDER ROAD HIGH SCHOOL

APRIL 2021

PHYSICAL SCIENCES CONTROL TEST 2

75 MINUTES

CO, JA, MH

GRADE 12

TOTAL = 60

Instructions:

- The question paper consists of 7 questions.
 - Answer all the questions.
 - Answer section A on the answer sheet provided AND section B on folio sheets.
 - A non-programmable calculator may be used.
 - Number the answers correctly according to the numbering system.
 - Round off to two (2) decimal places where necessary.
 - A formula sheet has been provided at the end of the question paper.
 - Half reaction Table 4B is provided on the back of the A5 answer sheet.
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SECTION A

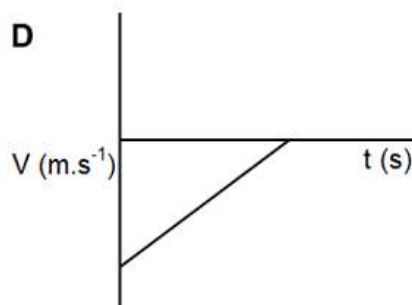
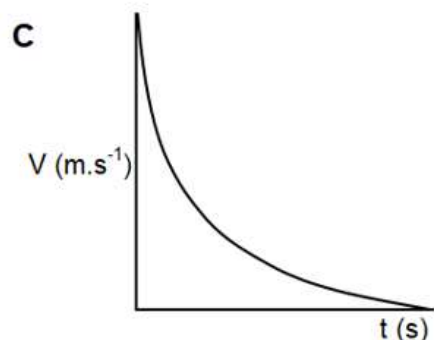
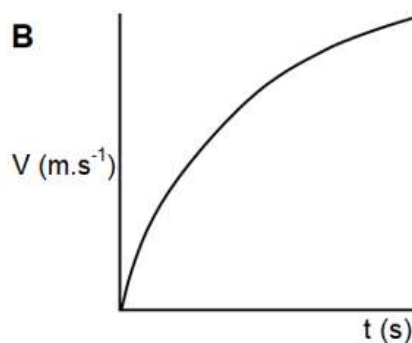
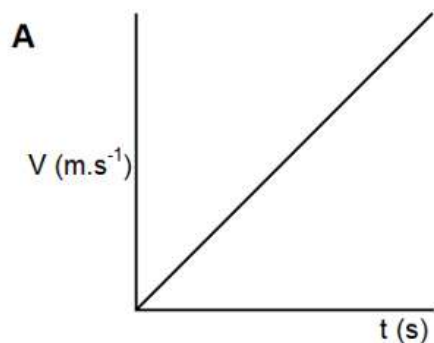
(answer on the answer sheet)

QUESTION 1:

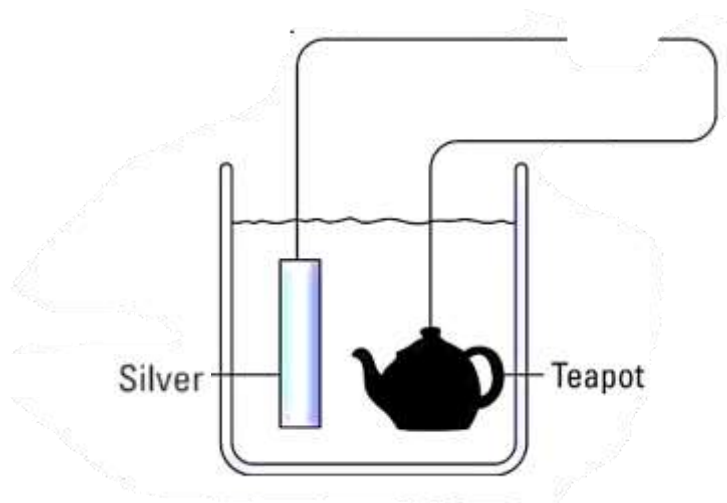
Four possible options are provided as answers to the following questions. Each question has only one correct answer. Choose the correct answer and write the letter (A – D) next to the relevant question number (1.1 – 1.3) on the answer sheet.

- 1.1 When an airbag inflates in a car during a collision, the chances of serious injury to a passenger is reduced because the ...
- A. passenger is brought to rest in a shorter period of time.
 - B. net force acting on the passenger is reduced.
 - C. passenger's change in momentum is reduced.
 - D. passenger's change in momentum is increased.

- 1.2 Which ONE of the following can be the velocity vs time graph of a projectile being thrown upwards?



- 1.3 A metal teapot needs to be electroplated with silver. Select the correct statement:



- A. The teapot is connected to the positive pole of the battery and reduction will take place on it, while the silver rod acts as the anode.
- B. The silver rod is reduced and the pot is connected to the negative pole and thus oxidised.
- C. The teapot is connected to the negative pole of the battery and reduction will take place on it, while the silver rod acts as the anode.
- D. The silver rod is oxidised and the pot is connected to the positive pole and thus reduced.

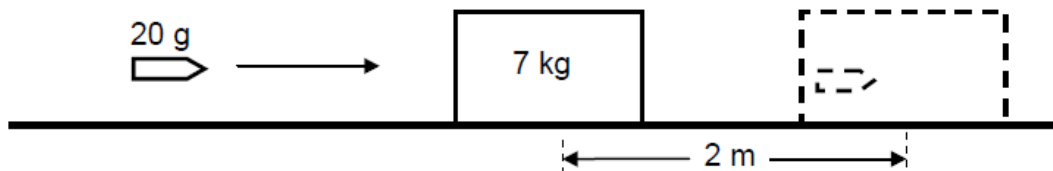
TOTAL SECTION A = [6]

SECTION B

(answer on folio paper)

QUESTION 2:

The diagram below shows a bullet of mass 20 g that is travelling horizontally. The bullet strikes a stationary 7 kg block with a velocity of $300 \text{ m}\cdot\text{s}^{-1}$ and becomes embedded in the block. The bullet and block together travel a horizontal distance of 2m before coming to a stop. Ignore air resistance.



- 2.1 State the *principle of conservation of linear momentum* in words. (2)
- 2.2 Calculate the velocity of the block after it was struck by the bullet. (4)
- 2.3 Use a calculation to show whether the collision was elastic or not. (4)

[10]

QUESTION 3:

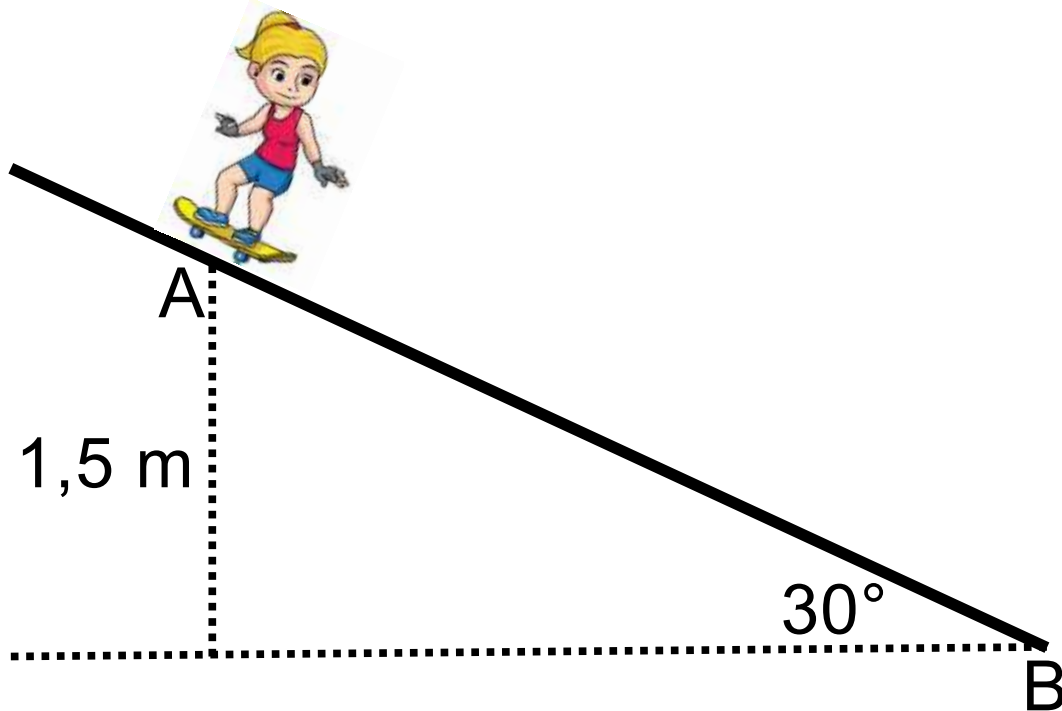
A man drops ball A from the edge of a window 45 m above the ground. One second later he throws a second ball, ball B, downwards with a velocity of $40 \text{ m}\cdot\text{s}^{-1}$. Ignore air friction.

- 3.1 Define the term *free fall*. (2)
- 3.2 Calculate how long after ball B was thrown, the two balls will cross. (5)

[7]

QUESTION 4:

An 80 kg skateboarder passes point A moving at $3 \text{ m}\cdot\text{s}^{-1}$ down the ramp inclined at 30° to the horizontal. Point A is 1,5 m above the ground and the force of friction between the ramp and the skateboard's wheels is 25 N.

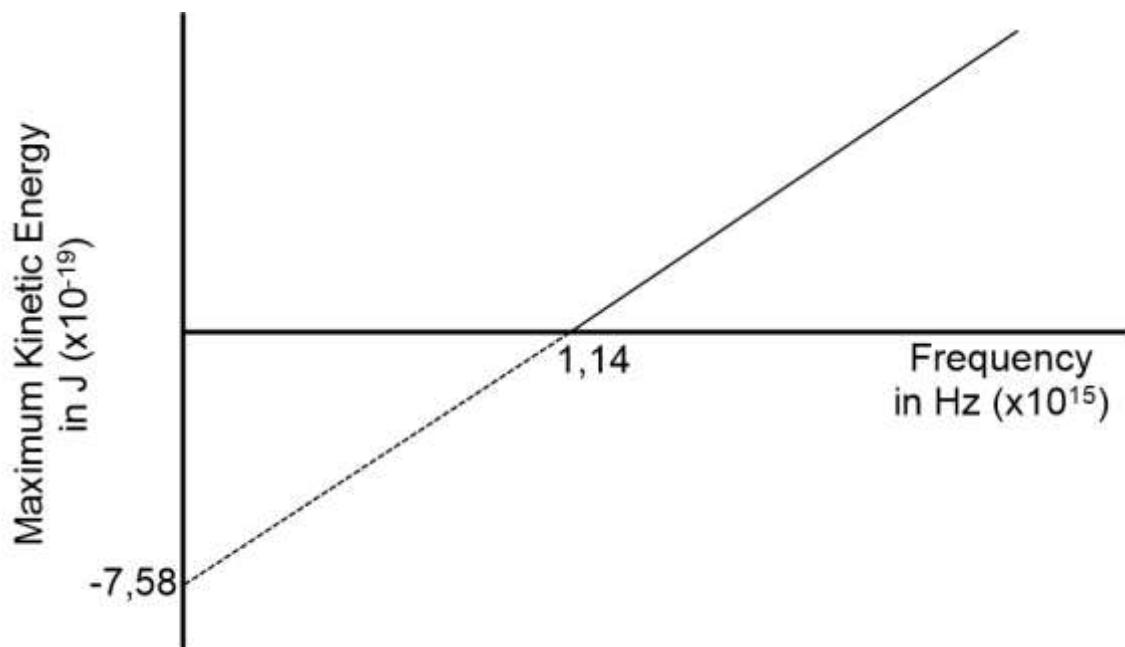


- 4.1 State the *work-energy theorem* in words. (2)
- 4.2 Use **energy principles** to calculate the speed of the skateboarder when she reaches the bottom of the ramp at point B. (5)
- 4.3 How will the speed of the skateboarder at point B change if the ramp is inclined at a larger angle? Write only INCREASES, DECREASES or REMAINS THE SAME. (1)

[8]

QUESTION 5:

The relationship between the frequency of light incident on a piece of copper metal and the maximum kinetic energy of ejected photoelectrons is investigated. The following graph, which is NOT drawn to scale, is obtained:



- 5.1 Define *work function*. (2)
- 5.2 Using the graph, write down the work function of copper. (1)
- 5.3 Calculate the maximum speed of a photoelectron when light of frequency $1,5 \times 10^{15}$ Hz is incident on the piece of copper metal. (3)
- 5.4 Would the maximum kinetic energy of the photoelectrons INCREASE, DECREASE or REMAIN THE SAME if the intensity of light mentioned in question 5.3 was increased? **Give a reason for your answer.** (2)

[8]

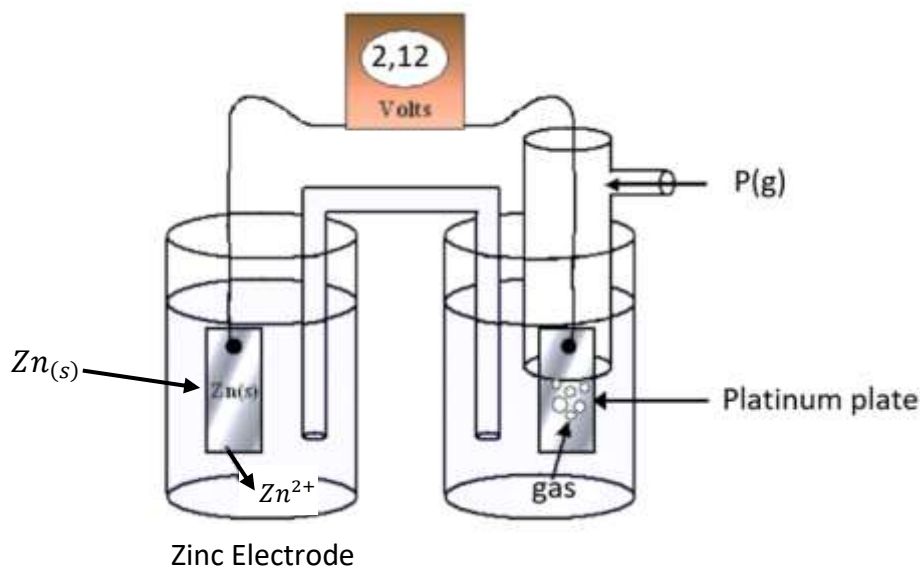
QUESTION 6:

- 6.1 A man standing on the side of the road hears the sound of a moving car's hooter at 96 Hz **higher** than what the driver of the car hears it. If the speed of the car is $30 \text{ m}\cdot\text{s}^{-1}$, and the observed frequency is 1076 Hz, calculate the speed of sound on the day. (4)
- 6.2 Name one medical application/instrument that uses the effect observed in 6.1. (1)

[5]

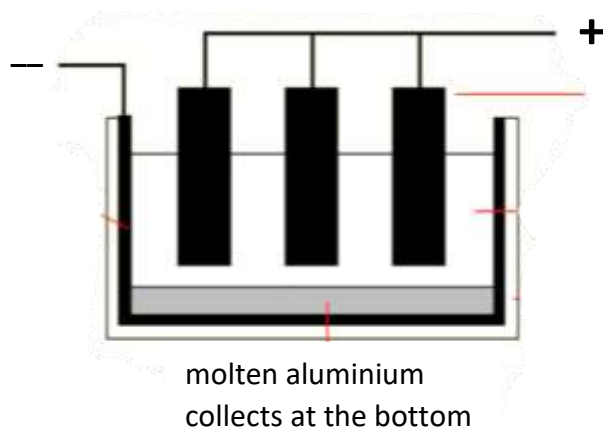
QUESTION 7:

7.1 The following diagram represents an electrochemical cell.



- 7.1.1 At which electrode will oxidation take place? (1)
- 7.1.2 Identify the substance P through calculation, if the initial voltage of the cell is 2,12 V. (4)
- 7.1.3 Explain, by using relative **reducing agent strength**, why this cell can produce energy. (2)
- 7.1.4 State one function of the salt bridge. (1)
- 7.1.5 Give the cell notation for this cell by using the answer of 7.1.3. (3)

7.2 The following cell is used to extract **aluminium** from its ore.



- 7.2.1 Define *electrolysis*. (2)
- 7.2.2 This cell is not spontaneous: what does this mean in terms of this cell? (1)
- 7.2.3 Give the reduction half reaction. (1)
- 7.2.4 Give one environmental disadvantage of this cell. (1)

[16]

TOTAL SECTION B = [54]

Formula Sheet

Physical Constants:

Name	Symbol	Value
Acceleration due to gravity	g	$9,8 \text{ m.s}^{-2}$
Speed of light in a vacuum	c	$3,0 \times 10^8 \text{ m.s}^{-1}$
Planck's constant	h	$6,63 \times 10^{-34} \text{ J.s}$
Charge on electron	e	$-1,6 \times 10^{-19} \text{ C}$
Electron mass	m_e	$9,11 \times 10^{-31} \text{ kg}$

Formulae:

MOTION

$v_f = v_i + a\Delta t$	$\Delta x = v_i\Delta t + \frac{1}{2}a\Delta t^2$ or $\Delta y = v_i\Delta t + \frac{1}{2}a\Delta t^2$
$v_f^2 = v_i^2 + 2a\Delta x$ or $v_f^2 = v_i^2 + 2a\Delta y$	$\Delta x = \left(\frac{v_f + v_i}{2}\right)\Delta t$ or $\Delta y = \left(\frac{v_f + v_i}{2}\right)\Delta t$

FORCE

$F_{\text{net}} = ma$	$w = mg$
$f_s^{\text{max}} = \mu_s N$	$f_k = \mu_k N$
$p = mv$	$F_{\text{net}}\Delta t = \Delta p$ $\Delta p = mv_f - mv_i$
$F = G\frac{m_1 m_2}{r^2}$	$g = G\frac{M}{r^2}$

WORK, ENERGY AND POWER

$W = F\Delta x \cos \theta$	$E_p = mgh$
$E_k = \frac{1}{2}mv^2$	$W_{\text{net}} = \Delta E_k$ $\Delta E_k = E_{k_f} - E_{k_i}$
$W_{\text{nc}} = \Delta E_k + \Delta E_p$	$P = \frac{W}{\Delta t}$
$P_{\text{ave}} = Fv_{\text{ave}}$	

WAVES, SOUND AND LIGHT

$v = f\lambda$	$T = \frac{1}{f}$
$f_L = \frac{v \pm v_L}{v \pm v_S} f_S$	$E = hf \quad \text{or} \quad E = \frac{hc}{\lambda}$
$E = W_0 + E_{k(\text{max})} \quad \text{where}$ $E = hf, W_0 = hf_0 \quad \text{and} \quad E_{k(\text{max})} = \frac{1}{2}mv_{\text{max}}^2$	

CHEMISTRY

$E_{\text{cell}}^{\theta} = E_{\text{cathode}}^{\theta} - E_{\text{anode}}^{\theta}$ or $E_{\text{cell}}^{\theta} = E_{\text{reduction}}^{\theta} - E_{\text{oxidation}}^{\theta}$ or $E_{\text{cell}}^{\theta} = E_{\text{oxidisingagent}}^{\theta} - E_{\text{reducingagent}}^{\theta}$

TABLE 4B: STANDARD REDUCTION POTENTIALS
 TABEL 4B: STANDAARD-REDUKSIEPOTENSIALE

Half-reactions/Halfreaksies	E^θ (V)
$\text{Li}^+ + e^- = \text{Li}$	-3,05
$\text{K}^+ + e^- = \text{K}$	-2,93
$\text{Cs}^+ + e^- = \text{Cs}$	-2,92
$\text{Ba}^{2+} + 2e^- = \text{Ba}$	-2,90
$\text{Sr}^{2+} + 2e^- = \text{Sr}$	-2,89
$\text{Ca}^{2+} + 2e^- = \text{Ca}$	-2,87
$\text{Na}^+ + e^- = \text{Na}$	-2,71
$\text{Mg}^{2+} + 2e^- = \text{Mg}$	-2,36
$\text{Al}^{3+} + 3e^- = \text{Al}$	-1,66
$\text{Mn}^{2+} + 2e^- = \text{Mn}$	-1,18
$\text{Cr}^{2+} + 2e^- = \text{Cr}$	-0,91
$2\text{H}_2\text{O} + 2e^- = \text{H}_2(\text{g}) + 2\text{OH}^-$	-0,83
$\text{Zn}^{2+} + 2e^- = \text{Zn}$	-0,76
$\text{Cr}^{3+} + 3e^- = \text{Cr}$	-0,74
$\text{Fe}^{2+} + 2e^- = \text{Fe}$	-0,44
$\text{Cr}^{3+} + e^- = \text{Cr}^{2+}$	-0,41
$\text{Cd}^{2+} + 2e^- = \text{Cd}$	-0,40
$\text{Co}^{2+} + 2e^- = \text{Co}$	-0,28
$\text{Ni}^{2+} + 2e^- = \text{Ni}$	-0,27
$\text{Sn}^{2+} + 2e^- = \text{Sn}$	-0,14
$\text{Pb}^{2+} + 2e^- = \text{Pb}$	-0,13
$\text{Fe}^{3+} + 3e^- = \text{Fe}$	-0,06
$2\text{H}^+ + 2e^- = \text{H}_2(\text{g})$	0,00
$\text{S} + 2\text{H}^+ + 2e^- = \text{H}_2\text{S}(\text{g})$	+0,14
$\text{Sn}^{4+} + 2e^- = \text{Sn}^{2+}$	+0,15
$\text{Cu}^{2+} + e^- = \text{Cu}^+$	+0,16
$\text{SO}_4^{2-} + 4\text{H}^+ + 2e^- = \text{SO}_2(\text{g}) + 2\text{H}_2\text{O}$	+0,17
$\text{Cu}^{2+} + 2e^- = \text{Cu}$	+0,34
$2\text{H}_2\text{O} + \text{O}_2 + 4e^- = 4\text{OH}^-$	+0,40
$\text{SO}_2 + 4\text{H}^+ + 4e^- = \text{S} + 2\text{H}_2\text{O}$	+0,45
$\text{Cu}^+ + e^- = \text{Cu}$	+0,52
$\text{I}_2 + 2e^- = 2\text{I}^-$	+0,54
$\text{O}_2(\text{g}) + 2\text{H}^+ + 2e^- = \text{H}_2\text{O}_2$	+0,68
$\text{Fe}^{3+} + e^- = \text{Fe}^{2+}$	+0,77
$\text{NO}_3^- + 2\text{H}^+ + e^- = \text{NO}_2(\text{g}) + \text{H}_2\text{O}$	+0,80
$\text{Ag}^+ + e^- = \text{Ag}$	+0,80
$\text{Hg}^{2+} + 2e^- = \text{Hg}(\text{l})$	+0,85
$\text{NO}_3^- + 4\text{H}^+ + 3e^- = \text{NO}(\text{g}) + 2\text{H}_2\text{O}$	+0,96
$\text{Br}_2(\text{l}) + 2e^- = 2\text{Br}^-$	+1,07
$\text{Pt}^{2+} + 2e^- = \text{Pt}$	+1,20
$\text{MnO}_2 + 4\text{H}^+ + 2e^- = \text{Mn}^{2+} + 2\text{H}_2\text{O}$	+1,23
$\text{O}_2(\text{g}) + 4\text{H}^+ + 4e^- = 2\text{H}_2\text{O}$	+1,23
$\text{Cr}_2\text{O}_7^{2-} + 14\text{H}^+ + 6e^- = 2\text{Cr}^{3+} + 7\text{H}_2\text{O}$	+1,33
$\text{Cl}_2(\text{g}) + 2e^- = 2\text{Cl}^-$	+1,36
$\text{MnO}_4^- + 8\text{H}^+ + 5e^- = \text{Mn}^{2+} + 4\text{H}_2\text{O}$	+1,51
$\text{H}_2\text{O}_2 + 2\text{H}^+ + 2e^- = 2\text{H}_2\text{O}$	+1,77
$\text{Co}^{3+} + e^- = \text{Co}^{2+}$	+1,81
$\text{F}_2(\text{g}) + 2e^- = 2\text{F}^-$	+2,87

Increasing oxidising ability/Toenemende oksiderende vermoë

Increasing reducing ability/Toenemende reduserende vermoë