12 Science March 2020 Test MEMO

- 1.1 D
- 1.2 D
- 1.3 C
- 1.4 D
- 1.5 B
- 2.1 In an isolated system, total linear momentum is conserved.
- 2.2 $\leq \rho_i = \leq \rho_f$ $m_i V_{i1} + m_2 V_{i2} = (m_i + m_2) V_f$ $\leftarrow \oplus$ $10(5) + (+0,05)(-250) = (10,05) V_f$ \vee $V_f = 3,73 \text{ m.s}^{-1} \text{ left}$
- 2.3 $E_k = \frac{1}{2} m_1 v_1^2 + \frac{1}{2} m_2 v_2^2$ = $\frac{1}{2} (0.05)(-250)^2 + \frac{1}{2} (10)(5)^2$ = 1687,5 J
 - E_{k} after $E_{k} = \frac{1}{2} mv^{2}$ = $\frac{1}{2} (10,05)(3,73)^{2}$ = 69,91 J

Ek before + Ex after : inclastic

- 3. Crift: $\Delta y = V_1 \Delta t + \frac{1}{2} \alpha \Delta t^2$ $\Delta y = 9.5(x) + \frac{1}{2}(-9.8)x^2$ $= 9.5(x) 4.9x^2$ Keys: $\Delta y = 0x + \frac{1}{2}(-9.8)x^2$ $-10 + \Delta y = -4.9x^2$ $\Delta y = 10 4.9x^2$
 - $9,5x-4,9x^{2} = 10-4,9x^{2}$ 9,5x = 10 x = 1,055
 - .. ay = 4,57 m
 - :. 4,57 m above the ground

4.2.1 What =
$$\Delta E_{k}$$

 $F_{net} \cdot \Delta z \cdot \cos \theta = \frac{1}{2} m v_{f}^{2} - \frac{1}{2} m v_{i}^{2}$ \[
\text{(147)} \left(\alpha \right) \left(\cos |80° \right) = \frac{1}{2} (75) \left(V_{F}^{2} \right) - \frac{1}{2} (75) \left(7)^{2} \right) \\
\tau \text{(147)} \left(\alpha \right) \left(\cos |80° \right) = \frac{1}{2} (75) \left(V_{F}^{2} \right) - \frac{1}{2} (75) \left(7)^{2} \right) \\
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4.2.2 Whet =
$$\Delta E_{K}$$
 (either)

Finet $\Delta x. \cos \theta = \frac{1}{2} m v_{f}^{2} - \frac{1}{2} m v_{i}^{2}$

$$[(75)(9,8)(\sin 40^{\circ}) + 147](x)(\cos 180^{\circ}) = \frac{1}{2}(75)(0)^{2} - \frac{1}{2}(75)(6,42)^{3}$$

[Finet = $F_{g_{11}} + f_{K}$]

5.1 The apparent change in the frequency of a wave as observed by an observer, because of relative motion between the source of the wave and the observer.

5.2
$$f_L = \frac{V^{\pm} V_L}{V^{\pm} V_S} f_S \checkmark$$

$$f_L = \underbrace{\frac{340 + (340 - 310)}{340}}_{= 304,71 \text{ Hz}} 280 \checkmark$$

- 6.1 Ag⁺ is a stronger oxidizing agent than Cu²⁺ and will be reduced to Ag ✓✓

 OR Cu is a stronger reducing agent than Ag and will be oxidized to Cu²⁺ (✓one mark only)
- 6.2.1 Chemical to electrical ✓
- 6.2.2 neutralise the charge imbalance in the solutions ✓

Or pathway for ions

OR completes the circuit

6.2.3
$$E^{\theta}_{cell} = E^{\theta}_{cathode} - E^{\theta}_{anode}$$

$$1.8\checkmark = 1.36 - E^{\Theta}_{anode}$$

$$E^{\Theta}_{anode} = -0.44V$$
Thus X = Fe \checkmark

- 6.2.4 X or iron or Fe ✓
- 6.3.1 copper sulphate or CuSO₄√

6.3.2 Cu
$$\rightarrow$$
 Cu²⁺ + 2e⁻ $\checkmark\checkmark$